

Fluid-rock interaction is decisive for the formation of tungsten deposits

Journal Article

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2 Fluid-rock interaction is decisive for the formation of

- 3 tungsten deposits
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15 ABSTRACT

Tungsten mineralization is typically associated with reduced granitic magmas of crustal origin. While this type of magmatism is widespread, economic tungsten deposits are highly localized with ~90% produced from only three countries worldwide. Therefore, the occurrence of reduced magmatism, while necessary for tungsten enrichment, seems to be insufficient to form such rare deposits. Here, we explore the mechanisms that lead to wolframite precipitation and evaluate whether they may exert a decisive control on tungsten global distribution. Tungsten differs from other rare metals

23 enriched in magmatic-hydrothermal ore deposits because it is transported as an anionic 24 species. Precipitation of the main tungstate minerals scheelite, CaWO₄, and wolframite, 25 (Fe,Mn)WO₄, thus depends on the availability of calcium, iron or manganese. We 26 demonstrate quantitatively that magmatic fluids at Panasqueira, Portugal, provide 27 tungsten in solution, whereas the host rock contributes the iron required to precipitate 28 wolframite. The combination of special source conditions with specific reactive host 29 rocks explains why major wolframite deposits are rare and confined to a few ore 30 provinces globally.

31 INTRODUCTION

32 Tungsten metal and tungsten alloys are characterized by toughness, corrosion resistance, and high strength at extreme temperatures. These physical properties make 33 34 tungsten essential for many industrial applications and, in combination with its localized 35 distribution, a *critical* metal and a high trade risk commodity (Klimek et al., 2015). 36 Despite previous studies outlining possible precipitation mechanisms (Heinrich, 1990; 37 Wood and Samson, 2000), indispensable controls on tungsten mineralization that 38 determine local and global resource distribution have not yet been identified for any 39 major deposit. Economic tungsten deposits represent >1000-fold enrichment in W from 40 background crustal levels of ~ 1 ppm (Rudnick and Gao, 2003) to ore grades of 1000– 41 50'000 ppm. The few global provinces hosting world-class tungsten deposits share 42 common magmatic-hydrothermal features. Deposits are typically associated with specific 43 magma types formed by melting of sedimentary rocks that may have been enriched in W 44 via weathering (Romer and Kroner, 2014, 2016); magmatic fractionation in the resulting 45 reduced melts causes selective enrichment of tungsten in the residual silicate melt

46 (Candela, 1992); saturation with alkali-rich magmatic volatiles results in tungsten transfer
47 to a hydrothermal fluid phase, (Manning and Henderson, 1984; Zajacz et al., 2008); and
48 finally precipitation of tungsten-rich minerals in sedimentary host rocks surrounding the
49 granites concentrates W to economic ore grades up to several weight percent as the most
50 significant enrichment step (Audétat et al., 2000).

51 The fundamental geochemical difference of tungsten compared to other rare 52 metals enriched in magmatic-hydrothermal ore deposits (e.g., porphyry Cu-Au or Sn vein 53 deposits) lies in the chemically 'hard' nature of metal – ligand interactions in aqueous solution (Pearson, 1963). Tungsten is transported as part of an oxy-anion, WO₄²⁻ that can 54 be complexed with Na⁺, K⁺ or H⁺ in hydrothermal fluids, in contrast to Cu⁺, Sn²⁺ or Au⁺ 55 56 that are transported as metal ions complexed with anionic ligands such as Cl- or HS-57 (Wood and Samson, 2000). Because of this particular chemical property, the precipitation 58 of the tungstate ore minerals requires that Fe, Mn or Ca are joined with W, but the 59 geological processes how the two components meet, and hence the essential controls on 60 tungsten ore deposition, are presently unknown.

61 **RESULTS**

To identify the primary controls on tungstate precipitation, fluid and rock chemistry have been analyzed at different spatial scales at a world-class wolframite vein deposit. Panasqueira (Portugal) is a giant tungsten deposit, one of the historic leading tungsten producers in Europe and one of the geologically best-characterized and bestpreserved tungsten vein deposits in the world (Kelly and Rye, 1979; Thadeu, 1951). Mineralization at Panasqueira consists of wolframite (dominantly FeWO₄) in subhorizontal cm- to m-scale quartz veins with lateral extents of hundreds of meters, hosted

69	in a vertically foliated biotite - chlorite schist (Foxford et al., 2000; Kelly and Rye, 1979).
70	The main oxide - silicate stage minerals in the veins include quartz + muscovite +
71	wolframite + arsenopyrite ± topaz. Ore veins cross-cut a hydrothermally altered granite
72	cupola and are asymmetrically disposed around it (Foxford et al., 1995).
73	Fluid inclusions associated with the main oxide - silicate stage in wolframite -
74	quartz veins at Panasqueira are CO ₂ -bearing water-rich liquids of uniform low salinity (5
75	- 8 wt% NaCleq). Laser-ablation inductively-coupled plasma mass spectrometry (LA-
76	ICPMS) using a high-sensitivity magnetic sector-field instrument (Wälle and Heinrich,
77	2014) was used to analyze W concentrations in the range of 1–70 ppm, together with
78	other elements including B, Fe, Mn (Fig. 1) Li, Cs and Rb (Appendix 2). While previous
79	studies have suggested that certain components such as He are mantle derived (Burnard
80	and Polya, 2004), fluids are systematically enriched in incompatible elements such as Li,
81	Cs and Rb, which is consistent with these components being derived from an evolved
82	magma. Importantly, fluids related with wolframite precipitation at Panasqueira
83	consistently show Fe and Mn concentrations below detection limit throughout the
84	deposit, even though the detection limits are as low as 0.3 (Mn) and 2 ppm (Fe) in the
85	best analyses provided by the largest inclusions (Fig. 1). These results show that the fluid
86	carried significant and variable concentrations of W, but insufficient Fe and Mn to
87	precipitate a significant fraction of the tungsten in the fluid alone. Wolframite deposition
88	therefore requires an external source of Fe. Multiple vein-type tungsten deposits show
89	chemical evidence of extensive fluid-rock interaction in tungsten vein type deposits
90	(Chicharro et al., 2016; Hulsbosch et al., 2015). Therefore, we focused our attention to
91	the host rock surrounding the veins.

92 The host rock at Panasqueira consists of vertically foliated biotite - chlorite schists 93 of the Beira group, which contain up to 7 wt % iron oxide and up to 0.1 wt % manganese 94 oxide (Polya, 1989). Trace metal enrichment of the schist occurs up to several kilometers 95 away from the altered granite cupola (Polya, 1988). At the deposit scale, the highest 96 tungsten grades are commonly associated with muscovite selvages growing into the veins 97 and with muscovite haloes in the host rock. Alteration is more extensive (dm- to m-scale) 98 in psammitic layers and spotted schists, compared to fine-grained schist and pelitic layers 99 (Foxford et al., 2000; Polya, 1989). The oxygen and sulfur isotopic signature is purely 100 magmatic previous to mineralization (Marignac, 1982, Polya et al., 2000), but the main 101 mineralizing stage signature lies between the magmatic fluid and host rock signature 102 (Kelly and Rye, 1979; Polya, 1989; Polya et al., 2000). Based on rock chemistry and the 103 evidence of fluid-rock interaction (Foxford, 1992; Kelly and Rye, 1979), the Beira schist 104 has the potential to contribute the cations required for tungstate precipitation. 105 To test whether reaction of the vein fluid with the wall rocks could provide the 106 missing Fe for wolframite precipitation, we studied the chemical and mineralogical 107 variation of the host rock as a function of distance from the vein. Decimeter-scale 108 samples of host rock contiguous to the vein wall were collected in schist (see appendix 1 109 for detailed methods). The schists proximal to the veins are systematically depleted in 110 iron, and enriched in potassium, aluminum, boron, tungsten and tin (Fig. 2). Iron 111 depletion in the regions proximal to the vein approach 40 wt% relative to sample 112 background in the spotted schist, and $\sim 5-10$ wt% in the fine grain schist. Note that the 113 elements enriched in the fluid (e.g., B, K, As, W) correspond systematically to those 114 enriched in the host rock (see Fig. 1). By contrast, elements absent in the fluid that are

115	common in the vein mineralogy (e.g., Fe) are systematically depleted in the host rock.
116	The chemical alteration in the host rock correlates also with changes in mineralogy. The
117	schist proximal to the vein is systematically depleted in chlorite and biotite, and enriched
118	in tourmaline, muscovite and arsenopyrite with the typical alteration mineralogy at
119	Panasqueira consisting of proximal early tourmaline and distal muscovite rich haloes
120	(Neiva, 2008) formed previous to coeval with mineralization (Marignac, 1982). Bleached
121	margins consisting dominantly of muscovite overprint the early alteration (Foxford, 1992
122	and references therein). Mass balance calculation indicates that even for relatively narrow
123	alteration haloes within the mine, the iron depletion from the host rock would yield
124	enough Fe to account for the main oxide - silicate stage mineralization (cut-off grade of
125	10 kg/m^2) in quartz veins at Panasqueira (Appendix 1 and 3). The lack of detectable iron
126	in the fluid is also consistent with iron being immediately removed from the fluid by
127	mineral precipitation.
128	DISCUSSION
129	Reactions of the fluid with the wall rock involve removal of Fe from the rock to
130	the fluid and addition of Al and B from the fluid to the rock. These chemical changes are
131	manifested at the mineralogical scale as (Fig. 3):
132	Muscovitization
133	Chlorite + 2 SiO ₂ + Al ³⁺ + K ⁺ = Muscovite + 2 Fe ²⁺ + H ₂ O
134	Biotite + 2 Al ³⁺ = Muscovite + 3 Fe^{2+}
125	

135 Tourmalinization

136 Chlorite + 1.4
$$SiO_2$$
 + 0.4 Al^{3+} + 0.4 Na^+ + 1.2 $B(OH)_3$

137 = $0.8 \text{ Fe}^{2+} + 3 \text{ H}_2\text{O} + 0.4 \text{ Tourmaline}$

138	Biotite + 1.4 Al ³⁺ + 0.4 Na ⁺ + 1.2 B(OH) ₃
139	= 0.4 Tourmaline + 1.8 Fe^{2+} + 0.6 SiO_2 + 2 H_2O + K^+
140	All these reactions liberate iron from the host rock into the fluid, and the
141	dominant chlorite-consuming reactions sequester alkalis from the fluid into the host rock.
142	Other muscovitization and tourmalinization reactions additionally lead to a pH increase
143	in the fluid. Note that, according to these reactions, the degree of enrichment/depletion of
144	different elements will depend on the chlorite/biotite content of the unaltered host rock,
145	as well as on the lateral extent of muscovitization and tourmalinization away from the
146	veins, which probably depends on rock permeability.
147	Precipitation of wolframite from an Fe-deficient ore fluid is driven by a threefold
148	effect of fluid-rock interaction on ferberite solubility according to
149	$Fe^{2+}(aq) + (Na,K)_2WO_4(aq) = FeWO_4$ (wolframite) +2 (Na,K) ⁺
150	$Fe^{2+}(aq) + H_2WO_4(aq) = FeWO_4$ (wolframite) +2 H ⁺
151	Alkali depletion and pH neutralization in the fluid due to tourmalinization and
152	muscovitization of the host rock decrease ferberite solubility. Iron enrichment in the fluid
153	increases the ferrous-tungstate ion activity product. Simultaneous occurrence of these
154	three processes thereby strongly promotes wolframite precipitation. The variability in the
155	element concentrations in the fluid is independent of distance from the greisen cupola
156	(especially for elements with low concentrations such as W, Fig. 1), and no systematic
157	change in fluid concentrations as a function of temperature (Appendix 3) has been found
158	during the tungsten mineralizing stage. These observations contrast, for example, with
159	copper concentrations in fluids in porphyry systems (Landtwing et al., 2005), wherein a
160	dramatic decrease in Cu concentration suggests chalcopyrite precipitation dominantly via

161 temperature decrease. Sudden pulses of pressure change may have contributed to 162 wolframite precipitation, but cannot precipitate ferberite unless iron is added to the 163 system. The spatially and temporally random variation in W concentration, the 164 homogeneous major composition, and the constant salinity of the mineralizing fluid 165 suggests that tungstate precipitation is not due to fluid mixing or boiling. More likely, it 166 reflects limited iron availability from variably permeable and reactive host rocks, which 167 controlled the slow growth of large wolframite crystals in the open vein space.

168 IMPLICATIONS FOR GLOBAL TUNGSTEN DISTRIBUTION

169 Globally, different W deposits show a correlation of tungstate mineralogy 170 (scheelite vs wolframite) with the nature of the wall rock (calcareous rocks vs iron-171 bearing schists and hornfels). The formation of scheelite deposits in skarns has previously 172 been attributed to a local Ca source (really, here a reference is needed!), but this is the 173 first study to demonstrate a host rock control on wolframite deposits. In particular, 174 tungsten at Panasqueira occurs dominantly as ferberite (FeWO₄) with only minor 175 hübnerite (MnWO₄; mole ratio of ~7:1 Fe:Mn). The predominance of iron tungstate over 176 manganese tungstate further reflects the host rock composition: the Beira schist rocks are 177 relatively iron rich but poor in manganese and calcium (Polya, 1989). Tungsten-vein 178 deposits worldwide show common features with Panasqueira: they consist of wolframite-179 bearing veins associated with greisenized granites, but in contrast to tin greisens are most 180 commonly not inside the granite but preferentially hosted in fine grained pelitic rocks 181 (e.g., Audétat et al., 2000, fig. 2). Globally, 70%–85% of deposits are hosted in shales 182 with various degree of contact metamorphism (Werner et al., 2014). Fractionated S-type 183 granites commonly exsolve fluorine and/or boron rich fluids during late crystallization

184 (Pollard et al., 1987). Due to the low solubility of fluorite (CaF₂) and the strong

185 partitioning of fluorine between fluid and ferromagnesian minerals, fluorine enrichment

is typically limited to late-stage magmatic fluids that are relatively poor in calcium and

187 iron (Barton, 1987). Scarcity of tourmaline in the granite despite the abundance of boron

188 in the system, and abundant schorl in the schist as a result of alteration further point to

189 low concentrations of iron in the magmatic fluid (Pivec et al., 1998). The consequence of

190 the scarcity of Fe and Ca is that a fluorine- and/or boron-rich magmatic fluids

191 characteristically derived from evolved granites, even if also rich in W, may be incapable

192 of ferberite/scheelite mineralization without significant input from external host rock

193 reactions. in tungsten vein deposits.

194 In summary, our study shows than not only scheelite deposits in Ca-rich rocks,

195 but also giant ferberite vein deposits in silicate rocks owe their existence and extreme

196 tungsten enrichment to reaction of fluids of magmatic origin with specific host rocks that

197 provide the necessary cations to precipitate tungsten from its anionic complex in

198 hydrothermal solutions.

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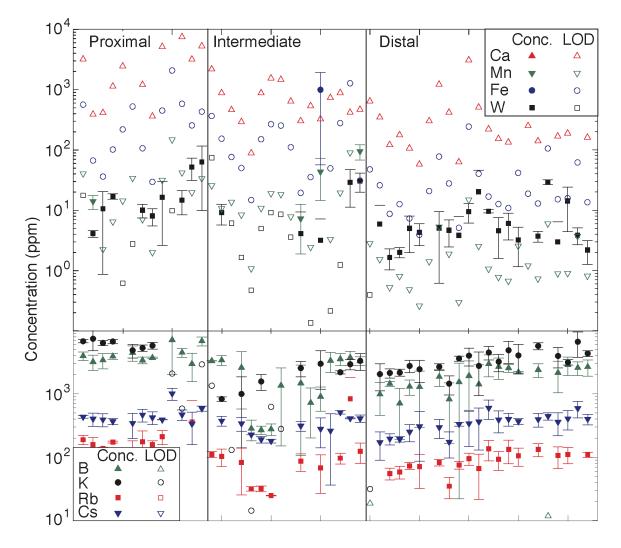
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305

306

308 FIGURE CAPTIONS





310 Figure 1. Average concentrations and standard deviations of assemblages of several

311 individual fluid inclusion (arranged in the *x*-axis). Fluid inclusion assemblages where the

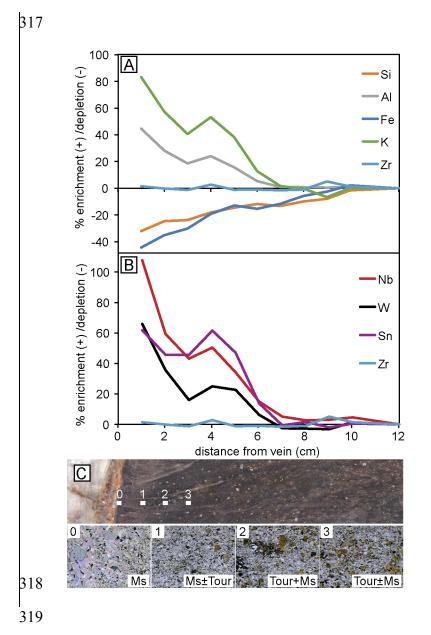
312 concentration is below detection limit are indicated by open symbols. "Proximal"

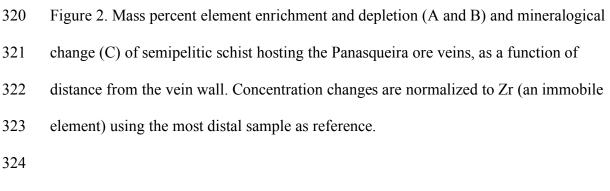
313 represents samples within <500 m from the greisen cupola at Panasqueira; "intermediate"

represents 500–1000 m from the cupola; and "distal" represents >1000 m from the

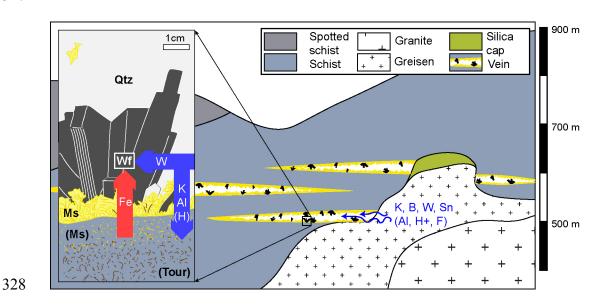
315 inferred magmatic source region.

316





327



329

330 Figure 3. Schematic representation of the fluid – rock reaction process leading to

331 wolframite (FeWO₄) mineralization at Panasqueira (Portugal), drafted on simplified cross

- 332 section from Thadeu (1951). Wf = wolframite, Ms = muscovite, Tour = tourmaline, and
- 333 Qtz = quartz.
- 334
- 335
- 336

¹GSA Data Repository item 2017xxx, xxxxxxxx, is available online at

- 338 <u>www.geosociety.org/datarepository/2017</u> or on request from <u>editing@geosociety.org</u>.
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